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[Indication of Fee]	
[Number of Pre-payment ledger]	043579
[Amount Paid]	¥21,000
[List of Documents attached]	
[Name of Item]	Specification 1
[Name of Item]	Abstract 1
[Number of Comprehensive Power of Attorney]	9102447
[Necessity of Proof]	Necessary

[Document Name] Specification

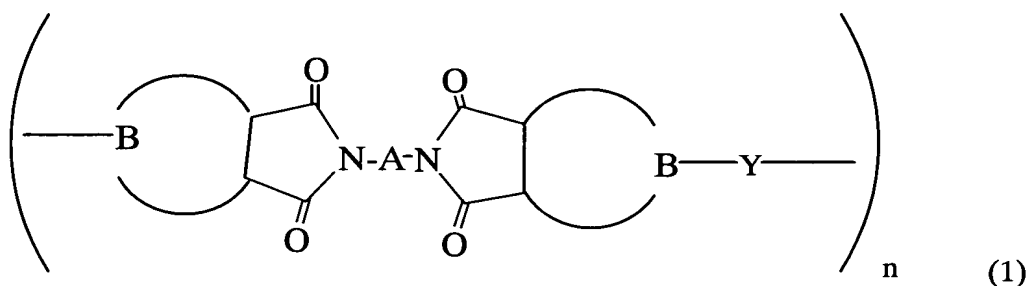
[Title of the Invention] Imide Silicone Resin and Production Process therefor

[Scope of Claim for Patent]

[Claim 1]

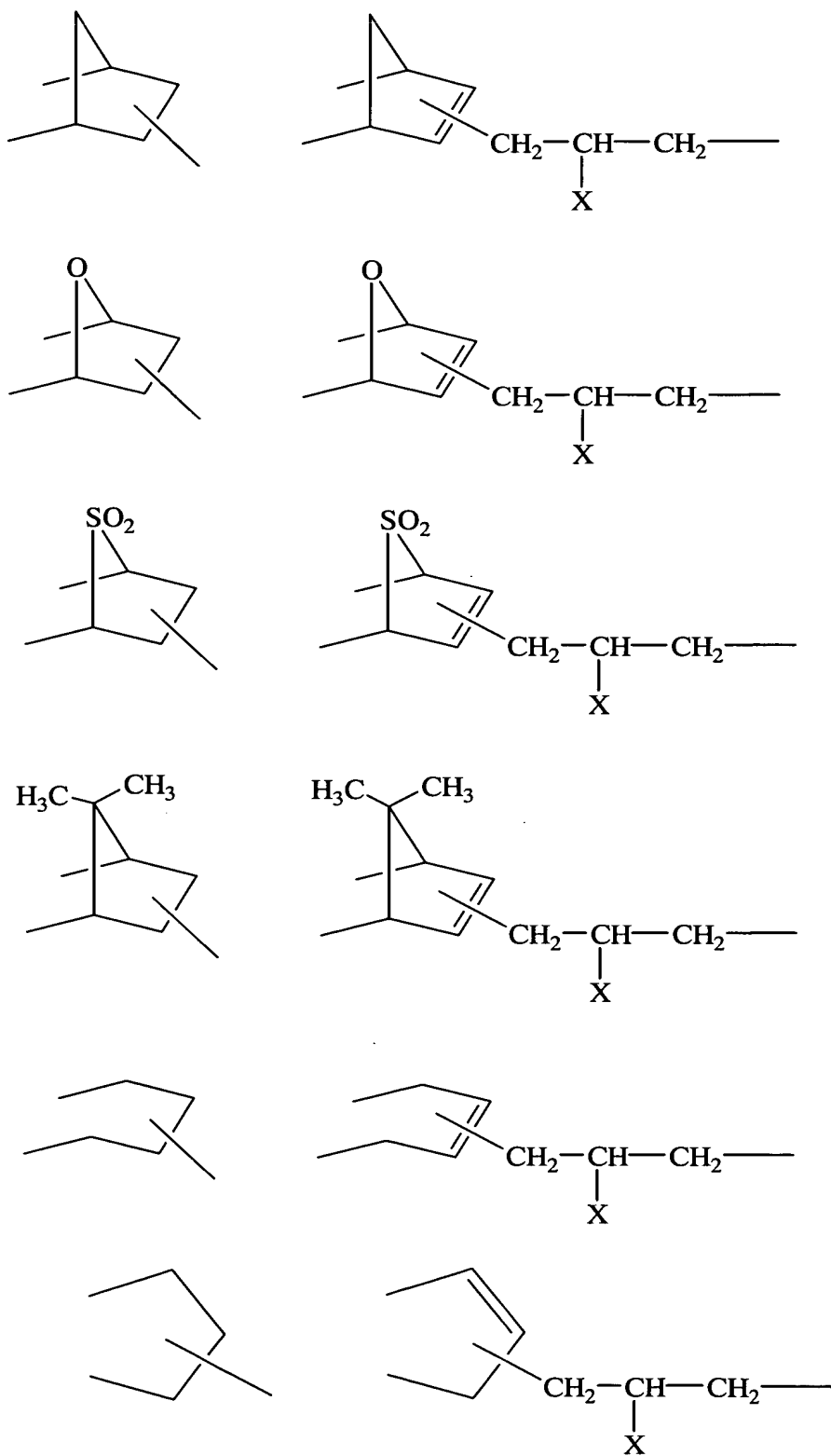
An imide silicone resin with a structure represented by a general formula (1) shown below:

[Chemical 1]



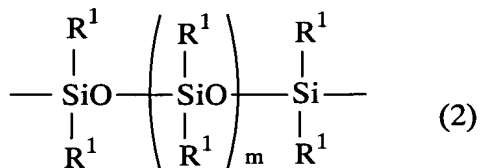
[wherein, A is a bivalent organic group, each B represents, independently, a trivalent group selected from groups shown below, in which two single bond sections protruding in a substantially identical direction are bonded to an imide ring to form a ring structure, Y is a bivalent group represented by a general formula (2) shown below, and n is an integer from 2 to 100:

[Chemical 2]



(wherein in each formula, X represents a hydrogen atom or a methyl group),

[Chemical 3]

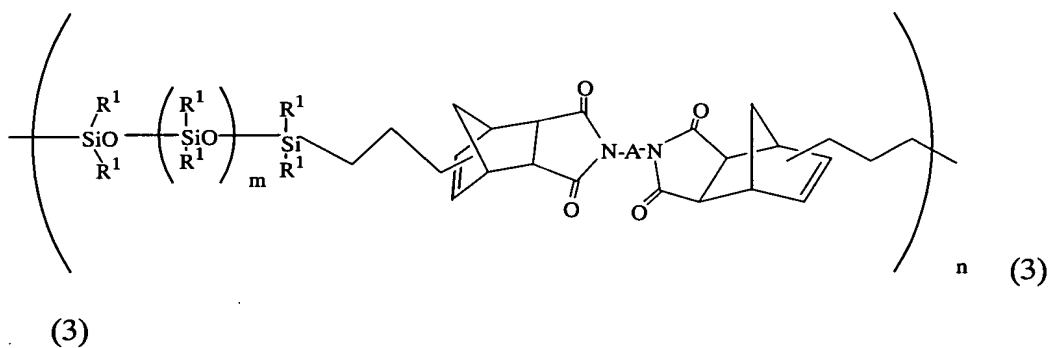


(wherein, each R^1 represents, independently, a monovalent organic group, and m is an integer from 0 to 100)].

[Claim 2]

The imide silicone resin defined in claim 1, with a structure represented by a general formula (3) shown below:

[Chemical 4]

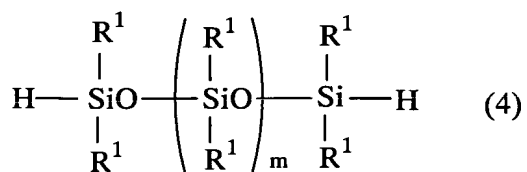


(wherein, R^1 , A, m , and n are as defined above).

[Claim 3]

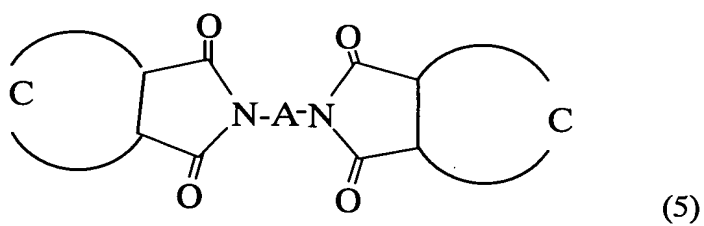
A production process for the imide silicone resin defined in claim 1, wherein an organopolysiloxane represented by a general formula (4) shown below and an imide compound represented by a general formula (5) shown below are subjected to an addition reaction:

[Chemical 5]



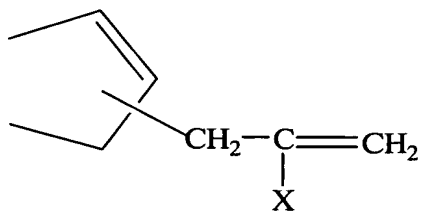
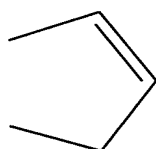
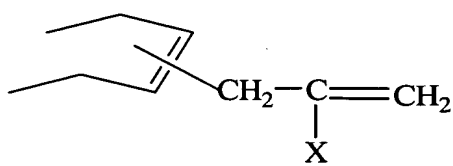
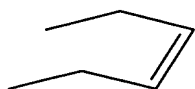
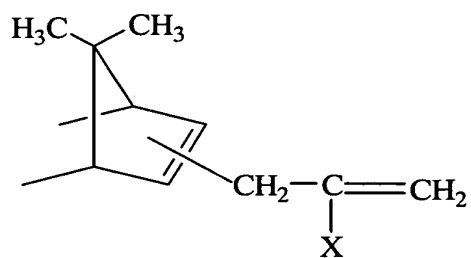
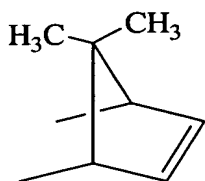
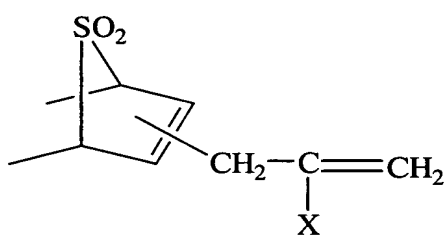
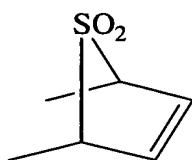
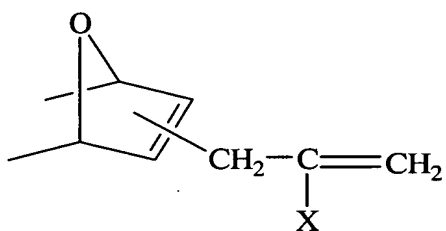
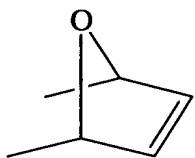
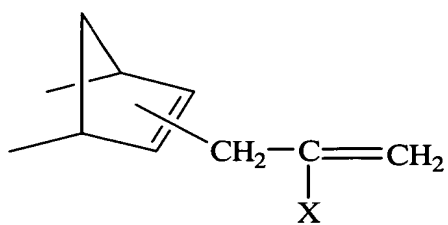
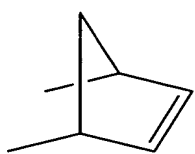
(wherein, each R¹ represents, independently, a monovalent organic group, and m is an integer from 0 to 100),

[Chemical 6]



[wherein, A is a bivalent organic group, and each C represents, independently, a bivalent group selected from groups shown below:

[Chemical 7]

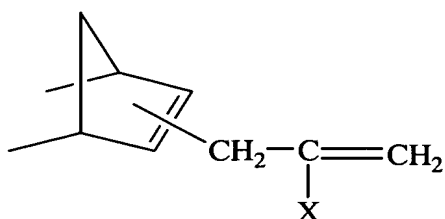


(wherein, X represents a hydrogen atom or a methyl group)].

[Claim 4]

The production process defined in claim 3, wherein in said imide compound represented by said general formula (5), said C in said general formula (5) is a bivalent group represented by a formula shown below:

[Chemical 8]



(wherein, X is as defined above).

[Claim 5]

A cured resin coating formed by curing an imide silicone resin defined in either one of claim 1 and claim 2.

[Detailed Description of the Invention]

[0001]

[Technical Field to Which the Invention belongs]

The present invention relates to a novel imide silicone resin that displays excellent heat resistance, solvent resistance, and adhesion and bonding to base materials, as well as a production process therefor, and a cured resin coating produced therefrom.

[0002]

[Prior Art]

Polyimide resins display excellent heat resistance and electrical insulation, and are consequently widely used as resin varnishes for electronic components and the like, and flexible printed wiring board materials. However, problems associated with

polyimide resins include poor flexibility because they are rigid, poor ease of use because they have high glass transition temperatures, and poor solubility in organic solvents. As a result, a variety of silicone modified polyimide resins have been proposed (for example, Laid-open publication (kokai) No. Hei 10-195278, Laid-open publication (kokai) No. Hei 8-34851). Silicone modified polyimide resins are able to compensate for the drawbacks of polyimide resins described above, while further improving the adhesion to base materials and the electrical characteristics.

[0003]

However, conventional syntheses of silicone modified polyimide resins suffer a number of problems, including requiring extreme synthesis conditions in that after an acid dianhydride and a diamine compound have been reacted together to synthesize a polyamic acid, a ring closing polyimidization reaction at a high temperature of at least 150°C is required, and the fact that the syntheses take considerable time. As a result, a resin material that offers at least the same functions as conventional silicone modified polyimide resins, is able to be synthesized more easily, and is also a thermosetting resin, has been required.

[0004]

[Problems the Invention Aims to Solve]

An object of the present invention is to provide a novel imide silicone resin capable of satisfying the above requirements.

[0005]

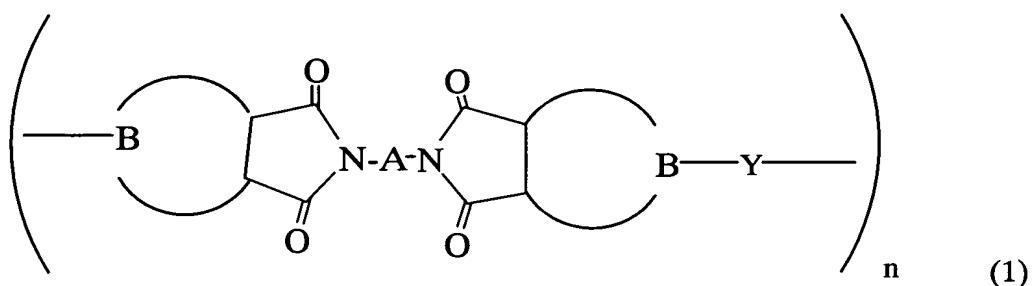
[Means for Solution of Problems]

As a result of intensive investigations aimed at achieving the above object, the inventors of the present invention finally arrived at the present invention. Namely, the

present invention provides an imide silicone resin with a structure represented by a general formula (1) shown below, and a production process therefor.

[0006]

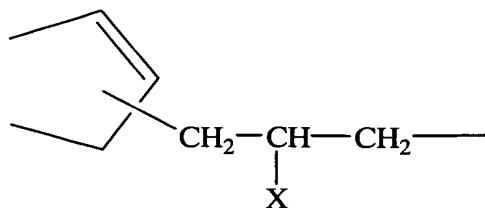
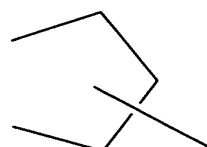
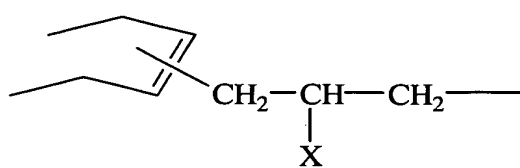
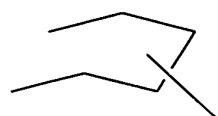
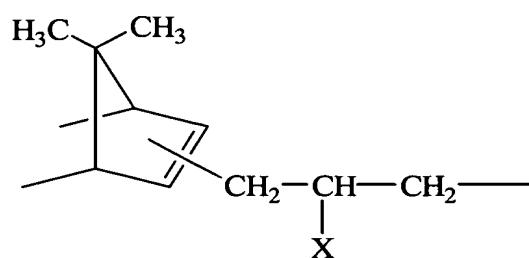
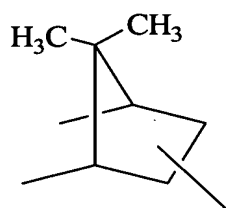
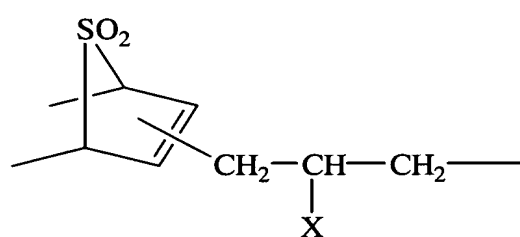
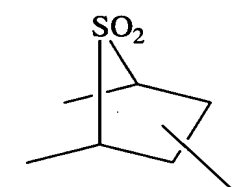
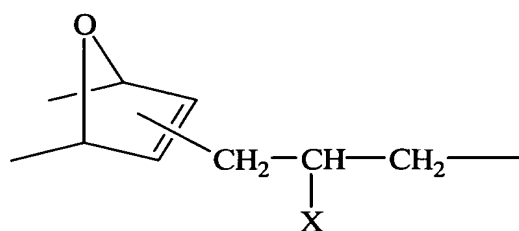
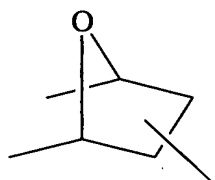
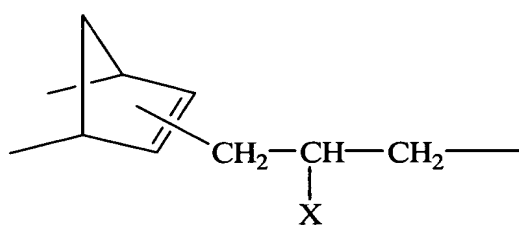
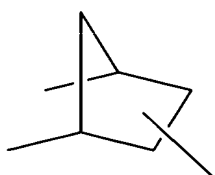
[Chemical 9]



[wherein, A is a bivalent organic group, each B represents, independently, a trivalent group selected from groups shown below, in which the two single bond sections protruding in a substantially identical direction are bonded to an imide ring to form a ring structure, Y is a bivalent group represented by a general formula (2) shown below, and n is an integer from 2 to 100.

[0007]

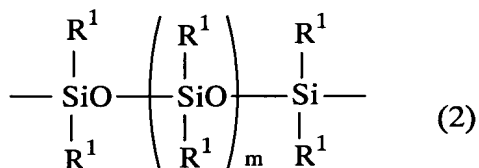
[Chemical 10]



(wherein in each formula, X represents a hydrogen atom or a methyl group)

[0008]

[Chemical 11]



(wherein, each R¹ represents, independently, a monovalent organic group, and m is an integer from 0 to 100)]

[0009]

[Embodiments of the Invention]

As follows is a detailed description of the present invention.

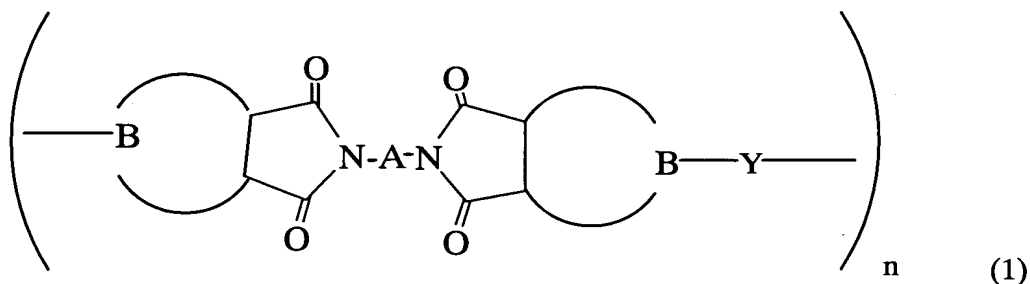
[0010]

[Imide Silicone Resin]

As described above, an imide silicone resin of the present invention has a structure represented by a general formula (1) shown below.

[0011]

[Chemical 12]

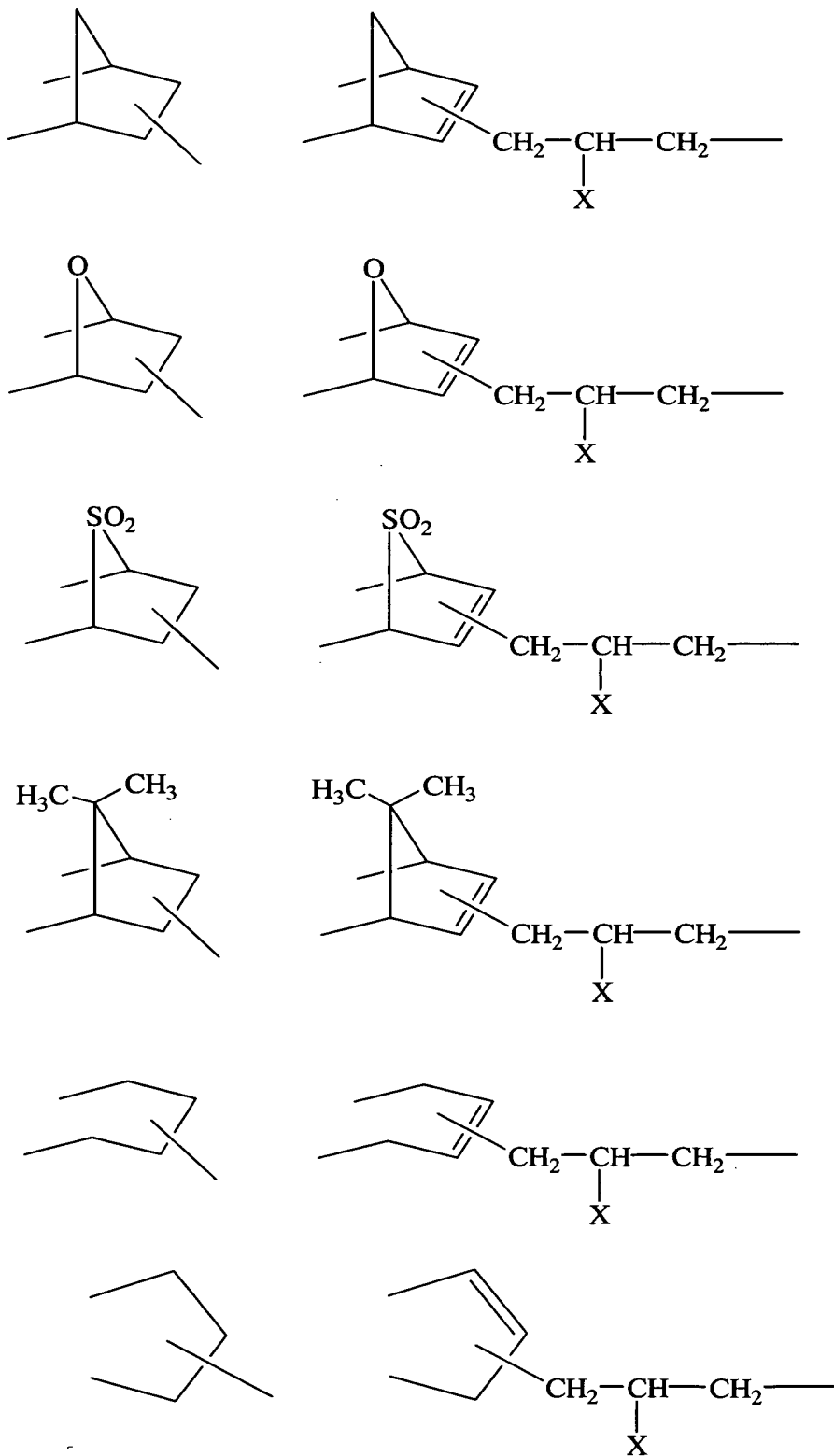


[wherein, A is a bivalent organic group, each B represents, independently, a trivalent group selected from groups shown below, in which the two single bond sections

protruding in a substantially identical direction are bonded to an imide ring to form a ring structure, Y is a bivalent group represented by a general formula (2) shown below, and n is an integer from 2 to 100, and preferably from 3 to 70.

[0012]

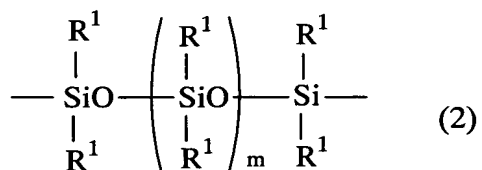
[Chemical 13]



(wherein in each formula, X represents a hydrogen atom or a methyl group)

[0013]

[Chemical 14]



(wherein, each R¹ represents, independently, a monovalent organic group, and m is an integer from 0 to 100, and preferably from 0 to 60)]

[0014]

Examples of the bivalent organic group A in the general formula (1) shown above include the groups shown below, although the bivalent organic group A is not restricted to these groups.

[0015]

groups such as cyclopentyl groups and cyclohexyl groups; aryl groups such as phenyl groups, tolyl groups, and xylyl groups; aralkyl groups such as benzyl groups and phenethyl groups; and halogenated alkyl groups such as 3,3,3-trifluoropropyl groups and 3-chloropropyl groups.

[0017]

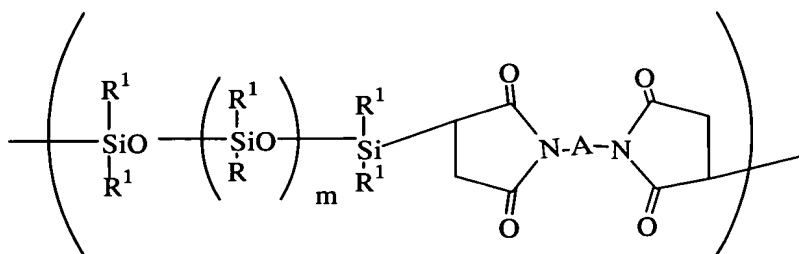
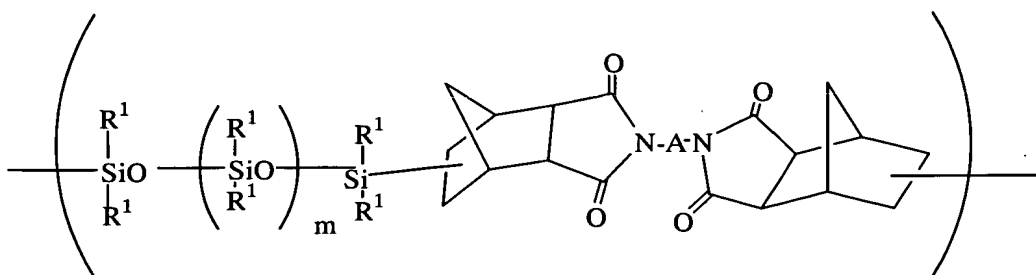
Furthermore, examples of the monovalent organic group R^1 in the general formula (2) shown above include unsubstituted or substituted monovalent hydrocarbon groups of 1 to 12 carbon atoms, and preferably of 1 to 8 carbon atoms, including alkyl groups such as methyl groups, ethyl groups, propyl groups, butyl groups, pentyl groups, and hexyl groups; cycloalkyl groups such as cyclopentyl groups and cyclohexyl groups; aryl groups such as phenyl groups, tolyl groups, and xylyl groups; aralkyl groups such as benzyl groups and phenethyl groups; halogenated alkyl groups such as 3,3,3-trifluoropropyl groups and 3-chloropropyl groups; trialkoxysilylated alkyl groups such as 2-(trimethoxysilyl)ethyl groups; as well as alkoxy groups such as methoxy groups, ethoxy groups, and propoxy groups; aryloxy groups such as phenoxy groups; and cyano groups.

[0018]

Examples of imide silicone resins of the present invention include resins with the repeating units shown in the formulas below, although the imide silicone resins are not restricted to them. Furthermore, copolymers containing 2 or more of the following repeating units are also possible.

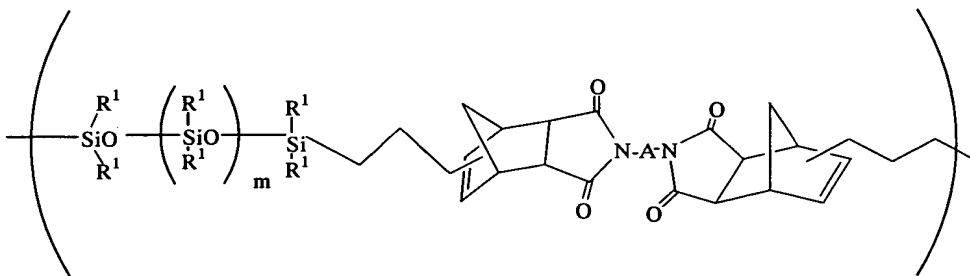
[0019]

[Chemical 16]



[0020]

[Chemical 17]



(wherein, A, R¹ and m are as defined above in relation to the general formula (1) and the general formula (2))

[0021]

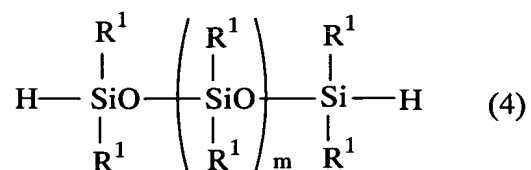
[Production of an Imide Silicone Resin]

<Organopolysiloxane>

In order to synthesize an imide silicone resin of the present invention, an organopolysiloxane represented by a general formula (4) shown below, with two hydrogen atoms bonded to the terminal silicon atoms, is used.

[0022]

[Chemical 18]



(wherein, R¹ is as defined above, and m is an integer from 0 to 100, and preferably from 0 to 60)

[0023]

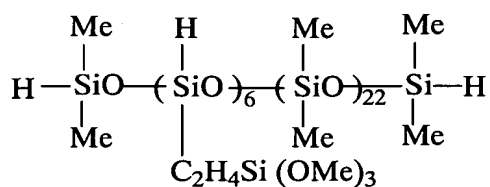
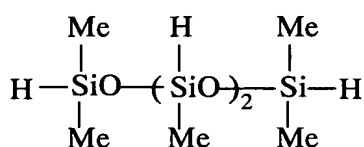
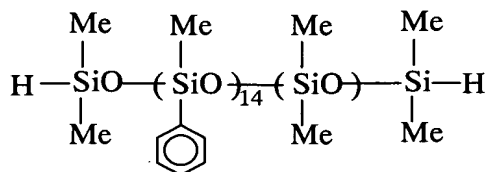
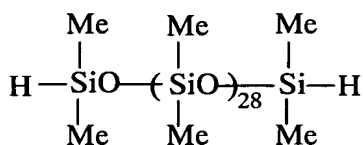
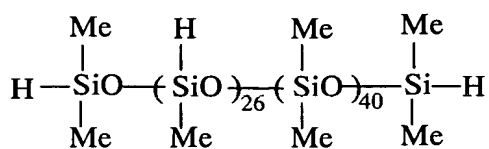
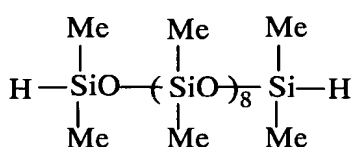
Examples of the organopolysiloxane described above include dimethylpolysiloxane with both molecular chain terminals blocked with dimethylhydrogensiloxy groups, copolymers of dimethylsiloxane and methylphenylsiloxane with both molecular chain terminals blocked with dimethylhydrogensiloxy groups, and methylphenylpolysiloxane with both molecular chain terminals blocked with dimethylhydrogensiloxy groups.

[0024]

Preferred specific examples are shown below, although the organopolysiloxane is not restricted to these structures (Me represents a methyl group).

[0025]

[Chemical 19]



These organopolysiloxanes can be used either singularly, or in combinations of two or more compounds.

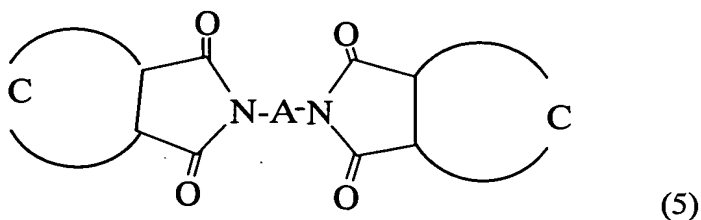
[0026]

<Imide Compound>

In order to synthesize an imide silicone resin of the present invention, an imide compound represented by a general formula (5) shown below, with two addition reaction reactive carbon-carbon double bonds, is used.

[0027]

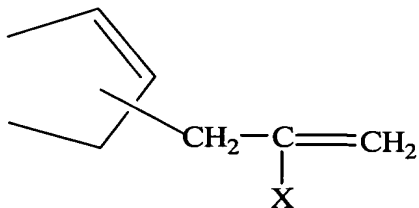
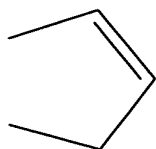
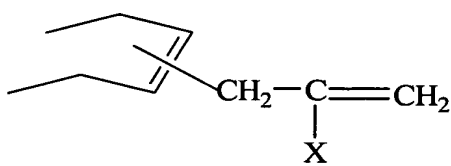
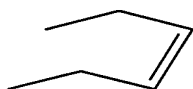
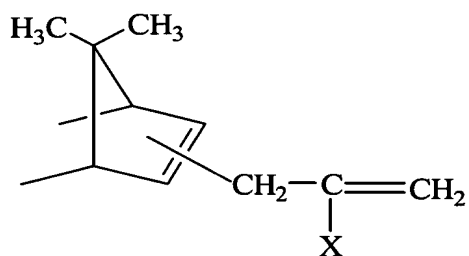
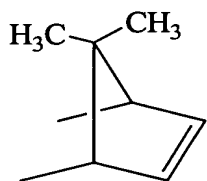
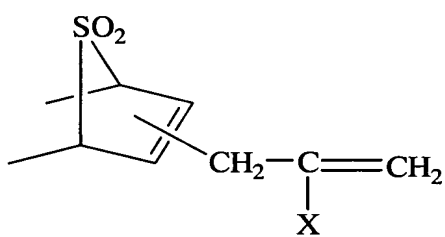
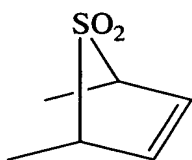
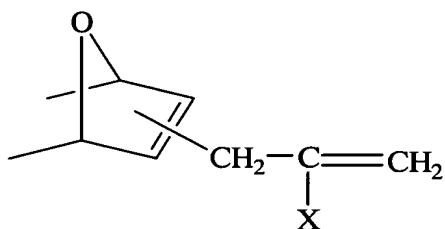
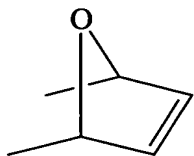
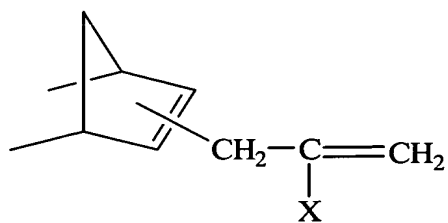
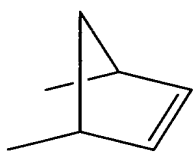
[Chemical 20]



[wherein, A is as defined above, and each C represents, independently, a bivalent group selected from the groups shown below:

[0028]

[Chemical 21]



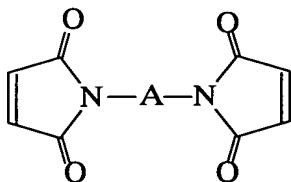
(wherein, X represents a hydrogen atom or a methyl group)].

[0029]

Furthermore, imide compounds represented by the general formula shown below can also be used.

[0030]

[Chemical 22]



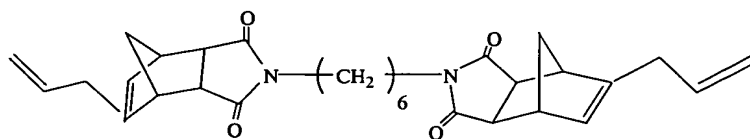
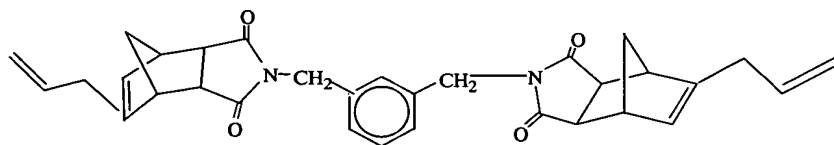
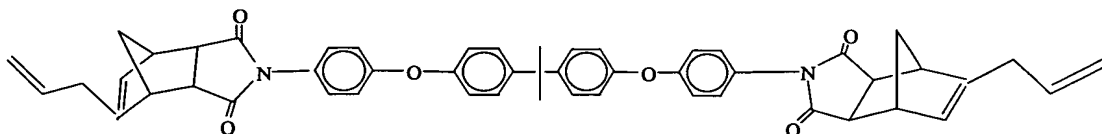
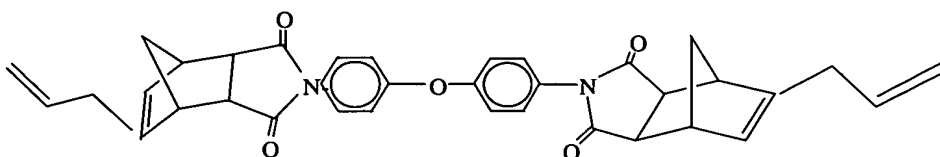
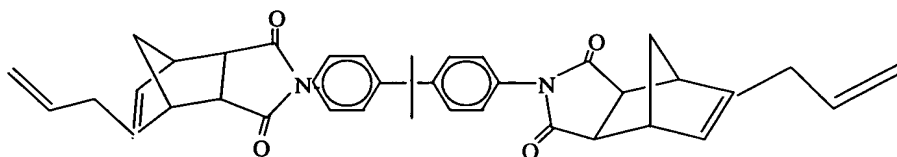
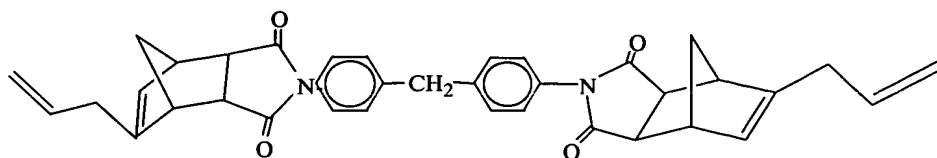
(wherein, A is as defined above)

[0031]

Specific examples of the imide compound used in the present invention are shown below, although the imide compound is not restricted to these compounds.

[0032]

[Chemical 23]



These imide compounds can be used either singularly, or in combinations of two or more compounds.

[0033]

In terms of the reactivity of the imide compound, in the case of an imide compound containing both an olefin based carbon-carbon double bond within a ring (in

other words, a bivalent group represented by $-\text{CH}=\text{CH}-$), and an olefin based carbon-carbon double bond such as an allyl group (a monovalent group such as $-\text{CH}=\text{CH}_2$) within a single molecule, the former (the bivalent group) contributes essentially nothing to the hydrosilylation reaction (addition reaction with a $\equiv\text{SiH}$ group) and is inactive, whereas the latter (the allyl group or the like) displays reactivity within the above reaction.

[0034]

<Addition Reaction>

A production process for an imide silicone resin according to the present invention involves an addition by hydrosilylation of an aforementioned imide compound containing an olefin based carbon-carbon double bond and an aforementioned organopolysiloxane (for example, an addition reaction of $\equiv\text{SiH}$ and either $\text{CH}_2=\text{CH}-$ or $-\text{CH}=\text{CH}-$), and the catalyst used during this reaction can be a conventionally known catalyst, with preferred catalysts including platinum based catalysts such as chloroplatinic acid, alcohol solutions of chloroplatinic acid, olefin complexes of platinum, alkenylsiloxane complexes of platinum, and carbonyl complexes of platinum; rhodium based catalysts such as tris(triphenylphosphine)rhodium; and iridium based catalysts such as bis(cyclooctadienyl)dichloroiridium. There are no particular restrictions on the quantity of the addition reaction catalyst used, which need only be an effective catalytic quantity, although a typical quantity is within a range from 0.001 to 20 parts by weight, and preferably from 0.01 to 5 parts by weight, per 100 parts by weight of the combined weight of the imide compound and the organopolysiloxane described above.

[0035]

Furthermore in the above addition reaction, a solvent may not be needed depending on the nature of the reaction raw materials described above, although use of a solvent where required is also acceptable. In those cases where a solvent is used, examples of suitable solvents include aromatic hydrocarbons such as benzene, toluene and xylene; ether based solvents such as tetrahydrofuran and ethylene glycol butyl ether acetate; aliphatic hydrocarbons such as hexane and methylcyclohexane; and polar solvents such as N-methyl-2-pyrrolidone, γ -butyrolactone and cyclohexanone.

[0036]

There are no particular restrictions on the reaction temperature, although the temperature is preferably within a range from 60°C to 120°C, and the reaction time is typically from approximately 30 minutes to approximately 12 hours.

[0037]

In addition, in the above addition reaction, if the $\equiv\text{SiH}$ group equivalence of the organopolysiloxane containing two or more hydrogen atoms bonded to silicon atoms (namely, $\equiv\text{SiH}$ groups) within one molecule is deemed α , and the equivalence of olefin based carbon-carbon double bonds that display reactivity in the hydrosilylation reaction within the imide compound containing two or more olefin based carbon-carbon double bonds within one molecule is deemed β , then the mixing ratio of the two components is typically $0.67 \leq \alpha/\beta \leq 1.5$, and preferably $0.95 \leq \alpha/\beta \leq 1.05$. If the value of the above ratio is outside this range, whether it be smaller or larger, then a large molecular weight cannot be expected, and a satisfactory cured resin coating cannot be formed from the produced imide silicone resin.

[0038]

[Imide Silicone Cured Resin Coating]

By dissolving an imide silicone resin of the present invention in a solvent such as those listed above as suitable for use within the addition reaction, including toluene, tetrahydrofuran or ethylene glycol butyl ether acetate, applying the solution to a substrate formed from a suitable base material, including glass, metals such as iron, copper, nickel or aluminum, or glass, volatilizing and removing the solvent to generate a film, and then heating at a temperature of 40°C to 400°C, and preferably of 80°C to 250°C, for a period of 0.01 to 30 hours, and preferably of 0.1 to 20 hours, a cured resin coating with a smooth surface, which displays excellent solvent resistance to alcohols, ketones and toluenes and the like, can be formed. The cured resin coating can be formed at any thickness within a range from 1 μm to approximately 1 cm, depending on the formation method used. Furthermore, the produced cured resin coating displays excellent adhesion and bonding to the base material.

[0039]

During the curing of an imide silicone resin of the present invention, a curing catalyst can also be added to accelerate the cross linking reaction. Examples of suitable curing catalysts include organic peroxides, onium salts and cation catalysts. Examples of organic peroxides include benzoyl peroxide, 2,4-diisopropyl benzene hydroperoxide, dicumyl peroxide, diisobutyl peroxide, bis-(4-t-butylcyclohexyl) peroxide, t-butylperoxy isobutyrate, 2,2'-azobis(2-methylbutyronitrile), dimethyl-2,2'-azobis(2-methylpropionate), and 2,2'-azobis[N-(2-propenyl)-2-methylpropioamide], examples of onium salts include pyridium p-toluenesulfonate, pyridinium m-nitrobenzenesulfonate, and benzyltriethylammonium chloride, and examples of cation catalysts include p-toluenesulfonic acid, methyl p-toluenesulfonate, and p-xylenesulfonic acid. If a

curing catalyst is used, then there are no particular restrictions on the quantity used, which need only be an effective catalytic quantity, although a typical quantity is within a range from 0.1 to 4 parts by weight per 100 parts by weight of the combined weight of the organopolysiloxane and the imide compound described above.

[0040]

Furthermore, the mechanism of the cross linking and curing reaction described above is thought to be based on cleavage of the olefin based carbon-carbon double bonds within the ring, which did not participate in the hydrosilylation reaction, and/or any terminal olefin based carbon-carbon double bonds remaining after the hydrosilylation reaction.

[0041]

[Other Components]

In addition, if required, inorganic fillers may also be added to an imide silicone resin of the present invention to form a resin composition. Examples of suitable inorganic fillers include fused silica, crystalline silica, alumina, carbon black, mica, clay, kaolin, glass beads, aluminum nitride, zinc oxide, calcium carbonate, and titanium oxide. These inorganic fillers can be used either singularly, or in combinations of two or more fillers. Furthermore, there are no particular restrictions on the quantity of the filler, although quantities from approximately 1 to approximately 500 parts by weight per 100 parts by weight of the imide silicone resin are preferred.

[0042]

Furthermore, if required, conductive particles may also be added to an imide silicone resin of the present invention to impart conductivity to the resin composition. Examples of suitable conductive particles include metal particles of gold, silver, copper

or nickel or the like, or particles in which a plastic or the like is covered with a metal. These conductive particles can be used either singularly, or in combinations of two or more particles. Furthermore, there are no particular restrictions on the quantity of the particles, although quantities from approximately 100 to approximately 1000 parts by weight per 100 parts by weight of the imide silicone resin are preferred.

[0043]

In addition, in order to improve the adhesion and bonding between the cured resin coating obtained by curing an imide silicone resin of the present invention and a base material, carbon functional silanes may also be added as required. Examples of carbon functional silanes include γ -glycidoxypropyltrimethoxysilane, γ -aminopropyltrimethoxysilane, 2-(γ -aminopropyl)ethyltrimethoxysilane, and vinyltrimethoxysilane. These compounds can be used either singularly, or in combinations of two or more compounds. Furthermore, typically quantities of the carbon functional silane are from approximately 0.1 to approximately 10 parts by weight per 100 parts by weight of the imide silicone resin.

[0044]

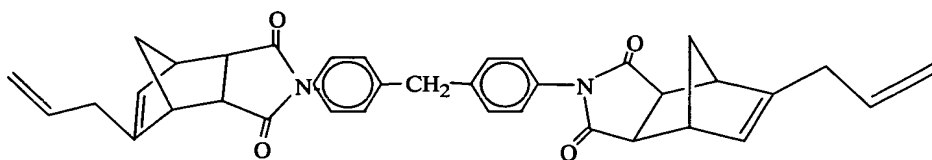
[Examples]

Example 1 (Synthesis of an Imide Silicone Resin)

In a 1 L flask equipped with a stirrer, a thermometer and a nitrogen exchange device were placed 100 parts by weight (0.175 mols) of an imide compound with olefin based carbon-carbon double bonds, as represented by a formula shown below:

[0045]

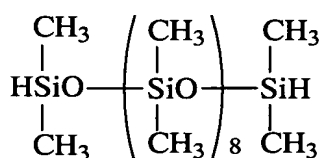
[Chemical 24]



128 parts by weight (0.176 mols) of an organopolysiloxane represented by an average structural formula shown below:

[0046]

[Chemical 25]



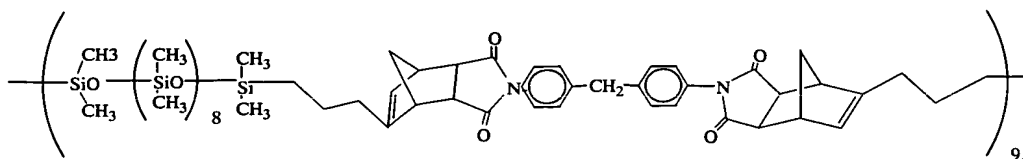
and 200 parts by weight of toluene, and subsequently 0.2 parts by weight of a 2% by weight ethanol solution of chloroplatinic acid was added, and the mixture was stirred for 5 hours at 90°C. The solvent was removed from the thus obtained product, which yielded 218 parts by weight of the target imide silicone resin. The external appearance of this imide silicone was of a light yellow colored, transparent solid. The number average molecular weight, as determined by gel permeation chromatography (GPC), was 12,000.

[0047]

As a result of GPC analysis, IR analysis, and ^1H -NMR analysis, it was evident that the produced imide silicone resin had a structure represented by the average structural formula shown below.

[0048]

[Chemical 26]



[0049]

The results of the IR analysis and the ^1H -NMR analysis are shown below.

- IR analysis

alkane C-H stretch: 2962 cm^{-1}

imide C=O stretch: 1778 cm^{-1} and 1714 cm^{-1}

imide C-N stretch: 1379 cm^{-1}

Si-C stretch: 1260 cm^{-1}

Si-O-Si stretch: 1099 cm^{-1}

- ^1H -NMR analysis (units: ppm)

Si- CH_3 : 0 to 0.3

Si- CH_2 : 0.4 to 0.6

imide double bond: 5.4 to 5.6

phenyl group of diphenylmethane: 5.7 to 6.4

Furthermore, peaks in the vicinity of 4.5 ppm, indicating the presence of SiH groups, were not detected.

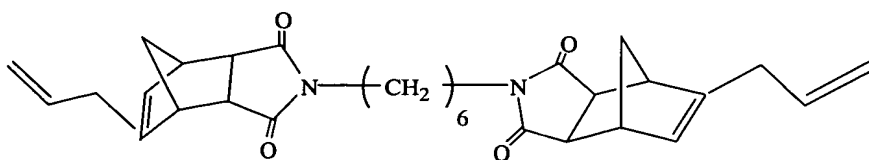
[0050]

Example 2 (Synthesis of an Imide Silicone Resin)

In a 5 L flask equipped with a stirrer, a thermometer and a nitrogen exchange device were placed 488 parts by weight (1.0 mol) of an imide compound with olefin based carbon-carbon double bonds, as represented by a formula shown below:

[0051]

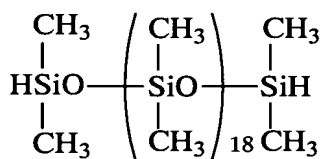
[Chemical 27]



1466 parts by weight (1.6 mols) of an organopolysiloxane represented by an average structural formula shown below:

[0052]

[Chemical 28]



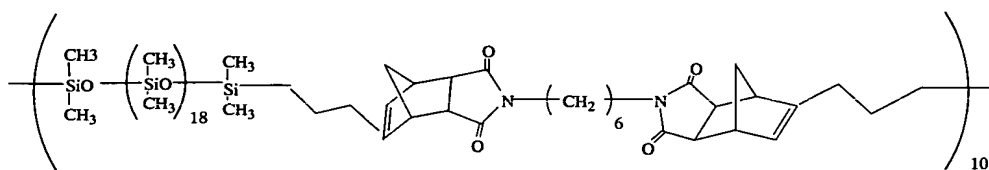
and 2000 parts by weight of toluene, and subsequently 0.8 parts by weight of a 2% by weight ethanol solution of chloroplatinic acid was added, and the mixture was stirred for 7 hours at 90°C. The solvent was removed from the thus obtained product, which yielded 1860 parts by weight of the target imide silicone resin. The external appearance of this imide silicone was of a light yellow colored, transparent solid with a tacky feel. The number average molecular weight, as determined by gel permeation chromatography (GPC), was 19,000.

[0053]

As a result of GPC analysis, IR analysis, and $^1\text{H-NMR}$ analysis, it was evident that the produced imide silicone resin had a structure represented by the average structural formula shown below.

[0054]

[Chemical 29]



[0055]

The results of the IR analysis and the ^1H -NMR analysis are shown below.

- IR analysis

alkane C-H stretch: 2963 cm^{-1}

imide C=O stretch: 1771 cm^{-1} and 1704 cm^{-1}

imide C-N stretch: 1379 cm^{-1}

Si-C stretch: 1260 cm^{-1}

Si-O-Si stretch: 1099 cm^{-1}

- ^1H -NMR analysis (units: ppm)

Si- CH_3 : 0 to 0.3

Si- CH_2 :-: 0.4 to 0.6

hexamethylene: 1 to 2

imide double bond: 5.6 to 5.9

Furthermore, peaks in the vicinity of 4.5 ppm, indicating the presence of SiH groups, were not detected.

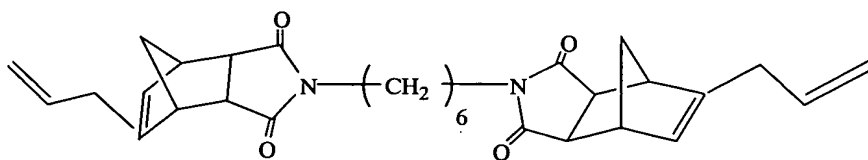
[0056]

Example 3 (Synthesis of an Imide Silicone Resin)

In a 2 L flask equipped with a stirrer, a thermometer and a nitrogen exchange device were placed 200 parts by weight (0.41 mols) of an imide compound with olefin based carbon-carbon double bonds, as represented by a formula shown below:

[0057]

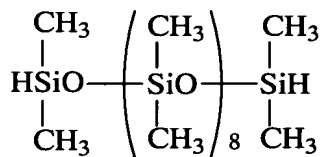
[Chemical 30]



298 parts by weight (0.41 mols) of an organopolysiloxane represented by an average structural formula shown below:

[0058]

[Chemical 31]



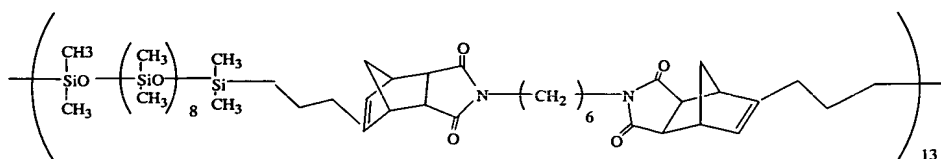
and 350 parts by weight of toluene, and subsequently 0.5 parts by weight of a 2% by weight ethanol solution of chloroplatinic acid was added, and the mixture was stirred for 5 hours at 90°C. The solvent was removed from the thus obtained product, which yielded 478 parts by weight of the target imide silicone resin. The external appearance of this imide silicone was of a light yellow colored, transparent viscous body. The number average molecular weight, as determined by gel permeation chromatography (GPC), was 16,000.

[0059]

As a result of GPC analysis, IR analysis, and ^1H -NMR analysis, it was evident that the produced imide silicone resin had a structure represented by the average structural formula shown below.

[0060]

[Chemical 32]



[0061]

The results of the IR analysis are shown below.

- IR analysis

alkane C-H stretch: 2963 cm^{-1}

imide C=O stretch: 1778 cm^{-1} and 1714 cm^{-1}

imide C-N stretch: 1379 cm^{-1}

Si-C stretch: 1260 cm^{-1}

Si-O-Si stretch: 1098 cm^{-1}

[0062]

Example 4 (Preparation of an Imide Silicone Cured Resin Coating)

Each of the imide silicone resins obtained in the examples 1 to 3 described above was dissolved in methyl ethyl ketone to form a resin solution with a concentration of 30% by weight.

Each of these resin solutions was applied to a glass substrate, and heated for 30 minutes at 60°C and then for 2 hours at 230°C , to prepare an imide silicone cured resin coating (thickness: $90\text{ }\mu\text{m}$).

[0063]

Each of the glass substrate bonded cured resin coatings obtained above was immersed for 5 minutes in methyl ethyl ketone at 25°C , and the surface of the coating was inspected for changes. The results are shown in Table 1.

The table entry "surface smooth" indicates that when the cured resin coating

surface was subjected to swelling with methyl ethyl ketone, no deformation or irregularities developed on the surface.

[0064]

Furthermore, each of the resin solutions was also applied to a copper substrate and a glass substrate, and heated for 30 minutes at 60°C and then for 2 hours at 230°C, to form an imide silicone cured resin coating (thickness: 15 μm) on each of the substrates. Subsequently, the coatings were left to stand for 72 hours in saturated water vapor at 2.1 atmospheres, and the cured coating on each substrate was subjected to a cross-hatch peeling test (JIS K 5400) to evaluate the adhesion following exposure to conditions of high humidity. The results are shown in Table 1.

[0065]

The numbers shown in Table 1 (numerator/denominator) represent the number of sections that peeled off (numerator) from the total number of sections 100 (denominator). In other words, a result of 100/100 indicates absolutely no peeling, whereas a result of 0/100 indicates that all the sections peeled off.

[0066]

Next, each of the resin solutions described above was applied to a copper substrate, and an imide silicone cured resin coating (thickness: 15 μm) was formed on the copper substrate under the same conditions as described above. Using these coated substrates as test pieces, the flexing properties were investigated using a 2mm ϕ mandrel. The results are shown in Table 1.

The symbol O represents good flexing properties, indicating that separation of the coating from the substrate, or cracking of the cured coating did not occur.

[0067]

[Table 1]

Resin	Solvent Resistance	Adhesion (cross-hatch peeling test)		Flexing properties
		Copper	Glass	
Example 1	surface smooth	100/100	100/100	O
Example 2	surface smooth	100/100	100/100	O
Example 3	surface smooth	100/100	100/100	O

[0068]

[Effects of the Invention]

A novel imide silicone resin of the present invention is capable of easily forming a cured resin coating by heat treatment. This cured resin coating displays excellent resistance to organic solvents such as ketones and the like, and also displays excellent adhesion and bonding to metal substrates such as copper, even under conditions of high humidity, as well as excellent durability. Accordingly, the imide silicone resin is useful for the surface protection of various metals, or as a protective material for semiconductor elements, a protective material for various substrates, an adhesive, or a heat resistant paint.

Furthermore, a production process for an imide silicone resin of the present invention is simple, and enables the target product to be obtained with good efficiency.

[Document Name] Abstract

[Abstract]

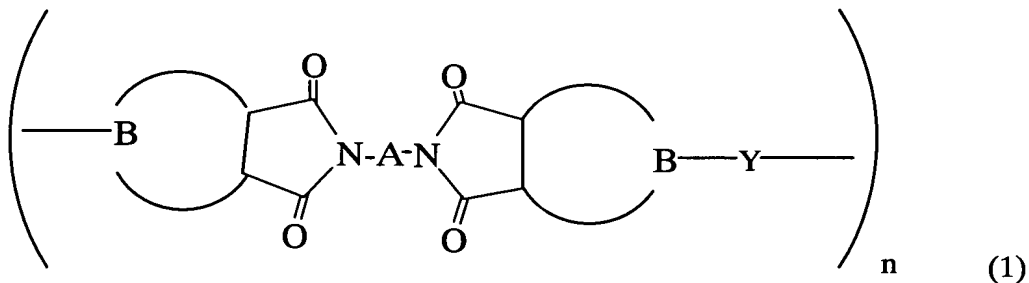
[Objects]

The provision of a novel imide silicone resin that on heat treatment is capable of easily forming a cured resin coating with excellent solvent resistance, humidity resistance and durability, as well as a production process therefor.

[Solution Means]

An imide silicone resin with a structure (1) shown below, which is the addition reaction product of an organopolysiloxane with two hydrogen atoms bonded to silicon atoms within each molecule, and an imide compound with two addition reaction reactive olefin based carbon-carbon double bonds within each molecule.

[Chemical 1]



[Selected Figure] None

Pat. 2002-259317

Approved or Additional Information

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ID Number	[000002060]
1. Date of Change:	22 August 1990
[Reason of Change]	New registration
Domicile	6-1, Otemachi 2-chome, Chiyoda-ku, Tokyo
Name	Shin-Etsu Chemical Co., Ltd.